

19. Numa notação devida ao físico P.A.M. Dirac, os estados próprios normalizados do oscilador harmónico numa dimensão são designados por $|\nu\rangle$. O estado fundamental é dado por $|0\rangle = \psi_0(x)$, onde $\psi_0(x)$ é dada por

$$\psi_0(\vec{r}) = \mathcal{N}_0 e^{-\frac{1}{2}\mu\omega r^2/\hbar} , \quad (1)$$

onde $\mathcal{N}_0$ representa a constante de normalização, $\mathcal{N}_0 = \left[\frac{\mu\omega}{\pi\hbar}\right]^{1/4}$

- a** Mostre que é igual a $\hbar\omega/2$ o valor próprio de $|0\rangle$ sob o efeito do Hamiltoniano dado no exercício (18) e igual a $\hbar\omega\left(\nu + \frac{1}{2}\right)$ o valor próprio de $(a^\dagger)^\nu |0\rangle$ (aplicar ν vezes o operador $a^\dagger$ definido no exercício 18).
- b** Na notação acima definida, o valor expectável de um operador A , caso o sistema se encontre no estado excitado $|\nu\rangle = \psi_\nu(x)$, é denotado por

$$\langle A \rangle = \langle \nu | A | \nu \rangle = \int_{-\infty}^{+\infty} dx \psi_\nu^*(x) A \psi_\nu(x) .$$

Determine o valor expectável do Hamiltoniano dado no exercício (18) para o estado $(a^\dagger)^\nu |0\rangle$.

- c** O estado $a^\dagger|\nu\rangle$ é proporcional ao estado $|\nu + 1\rangle$ (porquê?). Utilizando o resultado da alínea **b**, mostre que a constante de proporcionalidade é igual a $\sqrt{\nu + 1}$.
- d** Mostre que o valor expectável do operador x (exprimir em a e $a^\dagger$) para a transição de um estado $|\nu\rangle$ para um estado $|\nu + m\rangle$ se anula, salvo nos casos $m = \pm 1$.

Solution:

a. From the discussion regarding the solution of exercise (18d), we concluded that the eigenspectrum of H is given by $\frac{1}{2}\hbar\omega$, $\frac{3}{2}\hbar\omega$, $\frac{5}{2}\hbar\omega$, ..., whereas the respective eigenstates are given by $|0\rangle$, $|1\rangle$, $|2\rangle$, In a more compact notation

$$H|\nu\rangle = E_\nu|\nu\rangle \quad \text{with} \quad E_\nu = \left(\frac{1}{2} + \nu\right) \hbar\omega ,$$

b.

$$\langle H \rangle = \langle \nu | H | \nu \rangle = \int_{-\infty}^{+\infty} dx \psi_\nu^*(x) H \psi_\nu(x) =$$

$$\int_{-\infty}^{+\infty} dx \psi_\nu^*(x) E_\nu \psi_\nu(x) = E_\nu \int_{-\infty}^{+\infty} dx \psi_\nu^*(x) \psi_\nu(x) = E_\nu \int_{-\infty}^{+\infty} dx |\psi_\nu(x)|^2 = E_\nu 1 = E_\nu .$$

Hence, in the case that the system is in the eigenstate $|\nu\rangle$, E_ν is the value we expect to find when the total energy of the system, given by $\langle H \rangle$, is measured.

c. Using the result of exercise (18d), we obtain

$$H a^\dagger |\nu\rangle = \{E_\nu + \hbar\omega\} a^\dagger |\nu\rangle = \left\{ \left(\frac{1}{2} + \nu \right) \hbar\omega + \hbar\omega \right\} a^\dagger |\nu\rangle = \left\{ \frac{1}{2} + (\nu + 1) \right\} \hbar\omega a^\dagger |\nu\rangle = E_{\nu+1} a^\dagger |\nu\rangle \quad ,$$

whereas, for $|\nu + 1\rangle$ we also have $H|\nu + 1\rangle = E_{\nu+1}|\nu + 1\rangle$.

Now, since H has only one eigenstate for $E_{\nu+1}$, we must conclude that $a^\dagger |\nu\rangle$ and $|\nu + 1\rangle$ are proportional.

The proportionality constant, say $X_{\nu+1}$, can be found by determining

$$\begin{aligned} \int_{-\infty}^{+\infty} dx \left\{ a^\dagger \psi_\nu(x) \right\}^* a^\dagger \psi_\nu(x) &= \int_{-\infty}^{+\infty} dx \left\{ X_{\nu+1} \psi_{\nu+1}(x) \right\}^* X_{\nu+1} \psi_{\nu+1}(x) = \\ &= X_{\nu+1}^* X_{\nu+1} \int_{-\infty}^{+\infty} dx \psi_{\nu+1}^*(x) \psi_{\nu+1}(x) = |X_{\nu+1}|^2 \int_{-\infty}^{+\infty} dx |\psi_{\nu+1}(x)|^2 = |X_{\nu+1}|^2 \quad . \end{aligned}$$

So, we determine X_1 .

We start with ($\alpha = \mu\omega/\hbar$)

$$\begin{aligned} a^\dagger \psi_0 &= \sqrt{\frac{\mu\omega}{2\hbar}} \left(x - \frac{\hbar}{\mu\omega} \frac{\partial}{\partial x} \right) \mathcal{N}_0 e^{-\frac{1}{2}\alpha x^2} = \\ &= \mathcal{N}_0 \sqrt{\frac{\mu\omega}{2\hbar}} \left(x e^{-\frac{1}{2}\alpha x^2} - \frac{1}{\alpha} \frac{\partial}{\partial x} e^{-\frac{1}{2}\alpha x^2} \right) = \mathcal{N}_0 \sqrt{\frac{\mu\omega}{2\hbar}} \left(x e^{-\frac{1}{2}\alpha x^2} - \frac{1}{\alpha} \left\{ -\alpha x e^{-\frac{1}{2}\alpha x^2} \right\} \right) \\ &= \mathcal{N}_0 \sqrt{\frac{\mu\omega}{2\hbar}} \left(x e^{-\frac{1}{2}\alpha x^2} + x e^{-\frac{1}{2}\alpha x^2} \right) = 2\mathcal{N}_0 \sqrt{\frac{\mu\omega}{2\hbar}} x e^{-\frac{1}{2}\alpha x^2} \end{aligned}$$

So, we must then determine

$$|X_1|^2 = \int_{-\infty}^{+\infty} dx \left\{ a^\dagger \psi_0(x) \right\}^* a^\dagger \psi_0(x) = 2 |\mathcal{N}_0|^2 \frac{\mu\omega}{\hbar} \int_{-\infty}^{+\infty} dx x^2 e^{-\alpha x^2}$$

This integral may be looked up in the tables (Integral number 10 of the tables of integrals at <http://www.sosmath.com/tables/integral/integ37/integ37.html> maybe useful, note $\Gamma(n+1) = n\Gamma(n)$ en $\Gamma(1/2) = \sqrt{\pi}$.) We obtain then

$$|X_1|^2 = 2 \left[\frac{\mu\omega}{\pi\hbar} \right]^{1/2} \frac{\mu\omega}{\hbar} \frac{\frac{1}{2}\sqrt{\pi}}{(\mu\omega/\hbar)^{3/2}} = 1 = \sqrt{0+1} \quad .$$

This way, one can continue, for X_2 we must determine $a^\dagger \psi_1$ and its integral, etcetera. However, there exists a faster method.

First, we can make use of the general property

$$\int_{-\infty}^{+\infty} dx \left\{ A^\dagger \phi_1(x) \right\}^* \phi_2(x) = \int_{-\infty}^{+\infty} dx \phi_1^*(x) A \phi_2(x) \quad .$$

Then, we may substitute

$$aa^\dagger = aa^\dagger - a^\dagger a + a^\dagger a = [a, a^\dagger] + a^\dagger a = 1 + \left(\frac{H}{\hbar\omega} - \frac{1}{2}\right) = \frac{H}{\hbar\omega} + \frac{1}{2} \quad ,$$

whereas, moreover

$$\left(\frac{H}{\hbar\omega} + \frac{1}{2}\right) \psi_\nu = \left(\frac{E_\nu}{\hbar\omega} + \frac{1}{2}\right) \psi_\nu = \left(\frac{\left(\frac{1}{2} + \nu\right) \hbar\omega}{\hbar\omega} + \frac{1}{2}\right) \psi_\nu = (\nu + 1) \psi_\nu \quad .$$

This way, we obtain

$$\int_{-\infty}^{+\infty} dx \left\{ a^\dagger \psi_\nu(x) \right\}^* a^\dagger \psi_\nu(x) = \int_{-\infty}^{+\infty} dx \psi_\nu^*(x) aa^\dagger \psi_\nu(x) = \int_{-\infty}^{+\infty} dx \psi_\nu^*(x) (\nu + 1) \psi_\nu(x) = (\nu + 1).$$

Which gives $|X_{\nu+1}|^2 = \nu + 1$.

d.

$$a + a^\dagger = \frac{2\mu\omega}{\hbar} x \quad .$$

Hence,

$$\begin{aligned} \langle \nu | x | \nu + m \rangle &= \int_{-\infty}^{+\infty} dx \psi_\nu^*(x) \frac{\hbar}{2\mu\omega} (a + a^\dagger) \psi_{\nu+m}(x) = \\ &= \frac{\hbar}{2\mu\omega} \left\{ \int_{-\infty}^{+\infty} dx \psi_\nu^*(x) a \psi_{\nu+m}(x) + \int_{-\infty}^{+\infty} dx \psi_\nu^*(x) a^\dagger \psi_{\nu+m}(x) \right\} \\ &= \frac{\hbar}{2\mu\omega} \left\{ \int_{-\infty}^{+\infty} dx \left\{ a^\dagger \psi_\nu(x) \right\}^* \psi_{\nu+m}(x) + \int_{-\infty}^{+\infty} dx \psi_\nu^*(x) a^\dagger \psi_{\nu+m}(x) \right\} \\ &= \frac{\hbar}{2\mu\omega} \left\{ \int_{-\infty}^{+\infty} dx \{X_{\nu+1} \psi_{\nu+1}(x)\}^* \psi_{\nu+m}(x) + \int_{-\infty}^{+\infty} dx \psi_\nu^*(x) X_{\nu+m+1} \psi_{\nu+m+1}(x) \right\} \\ &= \frac{\hbar}{2\mu\omega} \left\{ X_{\nu+1}^* \int_{-\infty}^{+\infty} dx \psi_{\nu+1}^*(x) \psi_{\nu+m}(x) + X_{\nu+m+1} \int_{-\infty}^{+\infty} dx \psi_\nu^*(x) \psi_{\nu+m+1}(x) \right\} \\ &= \frac{\hbar}{2\mu\omega} \left\{ X_{\nu+1}^* \delta_{\nu+1, \nu+m} + X_{\nu+m+1} \delta_{\nu, \nu+m+1} \right\} \quad . \end{aligned}$$

The last step follows from the orthogonality relation of the normalized eigenfunctions of H , which reads

$$\int_{-\infty}^{+\infty} dx \psi_\nu^*(x) \psi_\mu(x) = \delta_{\nu, \mu} = \begin{cases} 1 & \text{for } \nu = \mu \\ 0 & \text{for } \nu \neq \mu \end{cases}$$

Now, $\delta_{\nu+1, \nu+m}$ is only non-zero for $\nu + 1 = \nu + m$ and in that case equal to 1, whereas $\delta_{\nu, \nu+m+1}$ is only non-zero for $\nu = \nu + m + 1$. In the former case $m = 1$, in the latter case $m = -1$.

Electromagnetic transitions of vibrational molecular states are related to the x operator. Hence, transitions are favoured when $\Delta\nu = \pm 1$.

20. Em virtude de se estar a estudar o espectro vibracional de uma molécula polar diatômica, considere que os dois núcleos atômicos estão ligados por uma mola (frequência de mola $\omega = 2\pi\nu$) e cujos movimentos estão restritos a uma dimensão (x).

a Utilizando os resultados do exercício (19), mostre que o espectro deste oscilador é dado por

$$E_v = \hbar\omega \left(v + \frac{1}{2} \right), \quad v = 0, 1, 2, 3, \dots$$

b A probabilidade de ocorrer uma transição dipolar entre os estados $|n\rangle$ e $|n+m\rangle$ é dada pelo integral $\int_{-\infty}^{+\infty} dx \psi_n^*(x) x \psi_{n+m}(x)$. Utilizando os resultados do exercício (19), determine as regras de selecção para transições dipolares.

c No entanto, o potencial entre os núcleos atômicos de uma molécula polar diatômica na realidade não é dado por um oscilador harmónico. O espectro real é dado por

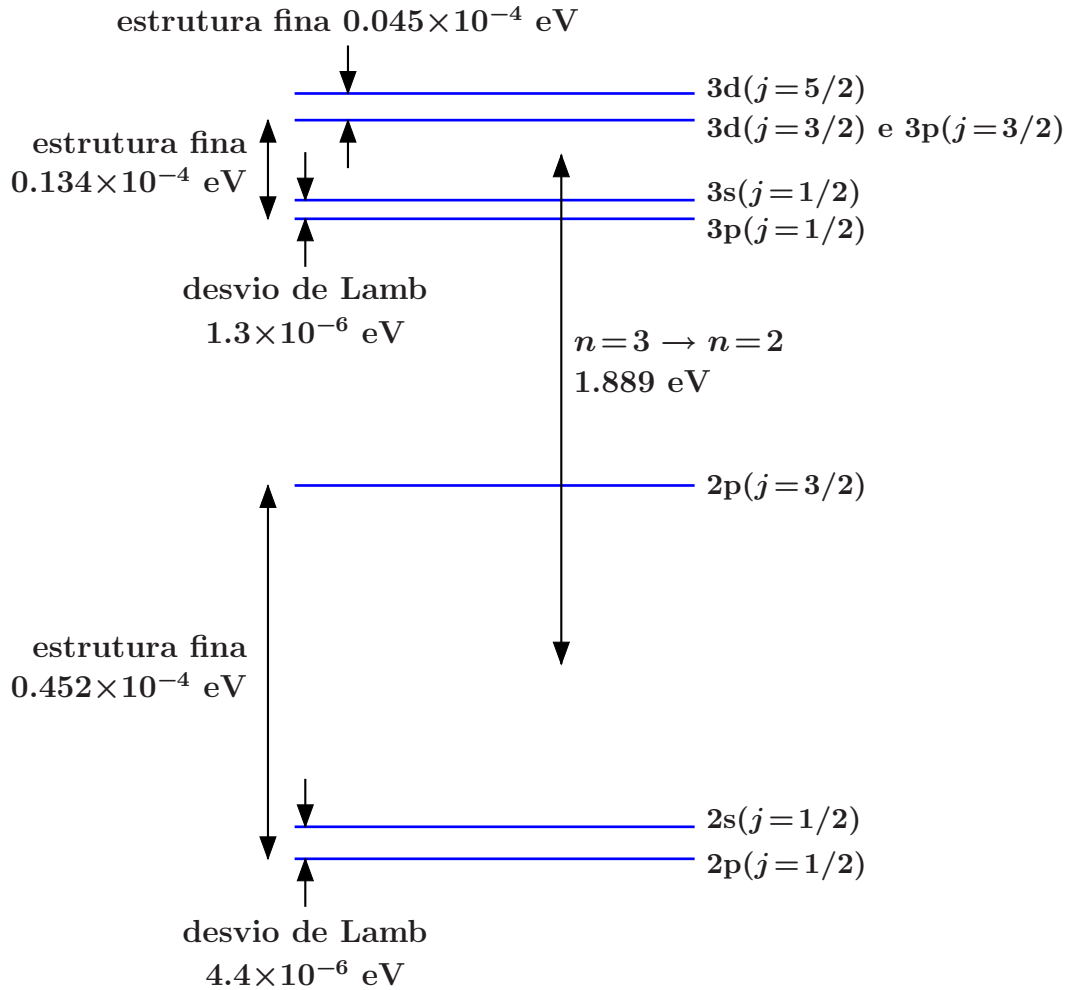
$$E_v = \hbar\omega \left(v + \frac{1}{2} \right) \left\{ 1 - a_1 \left(v + \frac{1}{2} \right) - a_2 \left(v + \frac{1}{2} \right)^2 - \dots \right\},$$

$$v = 0, 1, 2, 3, \dots,$$

onde os parâmetros $a_1, a_2, \dots$ variam de molécula para molécula. Também as regras de selecção são mais flexíveis na realidade, embora as transições $\Delta v = \pm 1$ sejam mais intensas do que as transições $\Delta v = \pm 2$, etc..

Numa aproximação linear ($a_2 = a_3 = \dots = 0$), determine $\hbar\omega$ e a_1 para a molécula HCl , caso o espectro de vibração de HCl mostre uma linha intensa com frequência 2886 cm^{-1} , uma linha mais fraca com frequência 5668 cm^{-1} e ainda uma linha muito fraca com frequência 8347 cm^{-1} .

21. Considere a primeira linha de Balmer ($n = 3 \rightarrow n = 2$). A figura mostra os níveis energéticos dos estados com $n = 2$ e $n = 3$ do átomo de hidrogénio. Além da estrutura fina do acoplamento spin-órbita está também apresentada a estrutura dos efeitos quânticos do electromagnetismo, o desvio de Lamb (prémio Nobel de 1955).



As várias transições electromagnéticas respeitam certas regras (as regras de selecção) devido às propriedades do campo electromagnético: O número quântico do momento angular ℓ apenas admite uma variação de uma unidade entre o estado inicial e o estado final (i.e. $\Delta\ell = \pm 1$), o número quântico do momento angular total j admite uma variação de uma unidade no máximo entre o estado inicial e o estado final (i.e. $\Delta j = 0, \pm 1$), enquanto o número quântico magnético j_z pode variar com $\Delta j_z = 0, \pm 1$ entre o estado inicial e o estado final.

- Explique os vários valores do número quântico j que aparecem na figura.
- Determine entre quais dos estados pode haver transições electromagnéticas e quantas linhas isto implica no espectro de hidrogénio.
- Caso a intensidade de uma linha depende linearmente do número de possibilidades que existam para o número quântico j_z , determine as razões das intensidades entre as várias linhas espectrais.

Solution:

We study here transitions $n = 3 \rightarrow n = 2$ for hydrogen. First, we will study how many states there are for $n = 3$ and $n = 2$ and then we will study how many dipolar transition can occur.

a. For $n = 3$ we have the possibilities $\ell < n = 3$, *i.e.* $\ell = 0$ (S), 1 (P) and 2 (D). Each combines with the two spin states $s_z = \pm 1/2$ of the electron.

For $\ell = 0$ we have one state $\ell_z = 0$. This combines to $j_z = \ell_z + s_z = 0 \pm 1/2 = \pm 1/2$. Hence, two states, which implies $j = 1/2$. Those states are collected under $3S_{\frac{1}{2}}$.

For $\ell = 1$ we have three states $\ell_z = 0, \pm 1$. Those combine to $j_z = \ell_z + s_z = 0 \pm 1/2 = \pm 1/2$, $j_z = \ell_z + s_z = -1 \pm 1/2 = -1/2$ and $-3/2$ and $j_z = \ell_z + s_z = +1 \pm 1/2 = +3/2$ and $+1/2$. Hence, six states in total, which can be grouped into two sets, one set of four states $\{j_z = -3/2, -1/2, +1/2, +3/2\}$, which implies $j = 3/2$, and one set of two states $\{j_z = -1/2, +1/2\}$, which implies $j = 1/2$. Those states are respectively collected under $3P_{\frac{3}{2}}$ and $3P_{\frac{1}{2}}$.

For $\ell = 2$ we have five states $\ell_z = 0, \pm 1, \pm 2$. Those combine to $j_z = \ell_z + s_z = 0 \pm 1/2 = \pm 1/2$, $j_z = \ell_z + s_z = -1 \pm 1/2 = -1/2$ and $-3/2$, $j_z = \ell_z + s_z = +1 \pm 1/2 = +3/2$ and $+1/2$, $j_z = \ell_z + s_z = -2 \pm 1/2 = -3/2$ and $-5/2$ and $j_z = \ell_z + s_z = +2 \pm 1/2 = +5/2$ and $+3/2$. Hence, ten states in total, which can be grouped into two sets, one set of four states $\{j_z = -5/2, -3/2, -1/2, +1/2, +3/2, +5/2\}$, which implies $j = 5/2$, and one set of four states $\{j_z = -3/2, -1/2, +1/2, +3/2\}$, which implies $j = 3/2$. Those states are respectively collected under $3D_{\frac{5}{2}}$ and $3D_{\frac{3}{2}}$.

level	n	ℓ	j	j_z
$3D_{\frac{5}{2}}$	3	2	$\frac{5}{2}$	$-5/2, -3/2, -1/2, +1/2, +3/2, +5/2$
$3D_{\frac{3}{2}}$	3	2	$\frac{3}{2}$	$-3/2, -1/2, +1/2, +3/2$
$3P_{\frac{3}{2}}$	3	1	$\frac{3}{2}$	$-3/2, -1/2, +1/2, +3/2$
$3P_{\frac{1}{2}}$	3	1	$\frac{1}{2}$	$-1/2, +1/2$
$3S_{\frac{1}{2}}$	3	0	$\frac{1}{2}$	$-1/2, +1/2$

For $n = 2$ we have the possibilities $\ell < n = 2$, *i.e.* $\ell = 0$ (S) and 1 (P). Each combines with the two spin states $s_z = \pm 1/2$ of the electron.

For $\ell = 0$ we have one state $\ell_z = 0$. This combines to $j_z = \ell_z + s_z = 0 \pm 1/2 = \pm 1/2$. Hence, two states, which implies $j = 1/2$. Those states are collected under $2S_{\frac{1}{2}}$.

For $\ell = 1$ we have three states $\ell_z = 0, \pm 1$. Those combine to $j_z = \ell_z + s_z = 0 \pm 1/2 = \pm 1/2$, $j_z = \ell_z + s_z = -1 \pm 1/2 = -1/2$ and $-3/2$ and $j_z = \ell_z + s_z = +1 \pm 1/2 = +3/2$ and $+1/2$. Hence, six states in total, which can be grouped into two sets, one set of four states $\{j_z = -3/2, -1/2, +1/2, +3/2\}$, which implies $j = 3/2$, and one set of two states $\{j_z = -1/2, +1/2\}$, which implies $j = 1/2$. Those states are respectively collected under $2P_{\frac{3}{2}}$ and $2P_{\frac{1}{2}}$.

level	n	ℓ	j	j_z
$2P_{\frac{3}{2}}$	2	1	$\frac{3}{2}$	$-3/2, -1/2, +1/2, +3/2$
$2P_{\frac{1}{2}}$	2	1	$\frac{1}{2}$	$-1/2, +1/2$
$2S_{\frac{1}{2}}$	2	0	$\frac{1}{2}$	$-1/2, +1/2$

b. Dipolar transitions $3 \rightarrow 2$ are limited by the selection rules $\Delta\ell = \pm 1$, $\Delta j = 0, \pm 1$ and $\Delta j_z = 0, \pm 1$. In the following table we collect the possible transitions.

	$2P_{\frac{3}{2}}$				$2P_{\frac{1}{2}}$		$2S_{\frac{1}{2}}$	
	$j_z = -\frac{3}{2}$	$j_z = -\frac{1}{2}$	$j_z = +\frac{1}{2}$	$j_z = +\frac{3}{2}$	$j_z = -\frac{1}{2}$	$j_z = +\frac{1}{2}$	$j_z = -\frac{1}{2}$	$j_z = +\frac{1}{2}$
$3D_{\frac{5}{2}} \quad j_z = -\frac{5}{2}$	X							
$3D_{\frac{5}{2}} \quad j_z = -\frac{3}{2}$	X	X						
$3D_{\frac{5}{2}} \quad j_z = -\frac{1}{2}$	X	X	X		$\Delta j = 2$		$\Delta j = 2$	
$3D_{\frac{5}{2}} \quad j_z = +\frac{1}{2}$		X	X	X			$\Delta \ell = 2$	
$3D_{\frac{5}{2}} \quad j_z = +\frac{3}{2}$			X	X				
$3D_{\frac{5}{2}} \quad j_z = +\frac{5}{2}$				X				
$3D_{\frac{3}{2}} \quad j_z = -\frac{3}{2}$	X	X			X			
$3D_{\frac{3}{2}} \quad j_z = -\frac{1}{2}$	X	X	X		X	X		
$3D_{\frac{3}{2}} \quad j_z = +\frac{1}{2}$		X	X	X	X	X	$\Delta \ell = 2$	
$3D_{\frac{3}{2}} \quad j_z = +\frac{3}{2}$			X	X		X		
$3P_{\frac{3}{2}} \quad j_z = -\frac{3}{2}$							X	
$3P_{\frac{3}{2}} \quad j_z = -\frac{1}{2}$							X	X
$3P_{\frac{3}{2}} \quad j_z = +\frac{1}{2}$		$\Delta \ell = 0$				$\Delta \ell = 0$	X	X
$3P_{\frac{3}{2}} \quad j_z = +\frac{3}{2}$								X
$3P_{\frac{1}{2}} \quad j_z = -\frac{1}{2}$							X	X
$3P_{\frac{1}{2}} \quad j_z = +\frac{1}{2}$		$\Delta \ell = 0$				$\Delta \ell = 0$	X	X
$3S_{\frac{1}{2}} \quad j_z = -\frac{1}{2}$	X	X	X		X	X		
$3S_{\frac{1}{2}} \quad j_z = +\frac{1}{2}$		X	X	X	X	X	$\Delta \ell = 0$	

From the table we observe that there are the seven possible transitions: $3D_{\frac{5}{2}} \rightarrow 2P_{\frac{3}{2}}$, $3D_{\frac{3}{2}} \rightarrow 2P_{\frac{3}{2}}$,

$3D_{\frac{3}{2}} \rightarrow 2P_{\frac{1}{2}}$, $3P_{\frac{3}{2}} \rightarrow 2S_{\frac{1}{2}}$, $3P_{\frac{1}{2}} \rightarrow 2S_{\frac{1}{2}}$, $3S_{\frac{1}{2}} \rightarrow 2P_{\frac{3}{2}}$ and $3S_{\frac{1}{2}} \rightarrow 2P_{\frac{1}{2}}$, represented by squares.

c. Inside the squares are indicated, with X , the various possibilities with respect to Δj_z .

transition	number of possibilities	relative intensity
$3D_{\frac{5}{2}} \rightarrow 2P_{\frac{3}{2}}$	12	1.0
$3D_{\frac{3}{2}} \rightarrow 2P_{\frac{3}{2}}$	10	$\frac{10}{12} = 0.83$
$3D_{\frac{3}{2}} \rightarrow 2P_{\frac{1}{2}}$	6	$\frac{6}{12} = 0.5$
$3P_{\frac{3}{2}} \rightarrow 2S_{\frac{1}{2}}$	6	$\frac{6}{12} = 0.5$
$3P_{\frac{1}{2}} \rightarrow 2S_{\frac{1}{2}}$	4	$\frac{4}{12} = 0.33$
$3S_{\frac{1}{2}} \rightarrow 2P_{\frac{3}{2}}$	6	$\frac{6}{12} = 0.5$
$3S_{\frac{1}{2}} \rightarrow 2P_{\frac{1}{2}}$	4	$\frac{4}{12} = 0.33$